CHAPTER FOUR

# Soil inorganic carbon, the other and equally important soil carbon pool: Distribution, controlling factors, and the impact of climate change

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#### Abstract

Soil inorganic carbon (SIC) contributes to up to half of the terrestrial C stock and is especially significant in arid and semi-arid environments, yet has not been explored as much as soil organic carbon (SOC). SIC plays an important role in agriculture, CO<sub>2</sub> sequestration and emission and climate regulation. To address this, a comprehensive review is presented on the digital mapping of soil inorganic carbon including a discussion of SIC vertical variation, its controlling factors, and sequestration/emission capability. We surveyed SIC distribution and mapping efforts in Australia, South Africa, Chile, the Mediterranean basin, Iran, China, France, and the United States. We found that current detailed spatial information on SIC distribution and stock is relatively scarce and digital soil mapping (DSM) efforts to address this are modest. Furthermore, we do not have a complete soil C model that explicitly accounts for all sources and sinks of soil carbon. This review showed that many aspects of SIC in DSM and soil C studies have been so far ignored and that SIC has a crucial role in climate regulation. This review provides some insights into the importance and unknown aspects of SIC.

# 1. Introduction

Soil carbon is recognized as central to soil functioning, plays a crucial role in maintaining the sustainability of soil to produce food, and acts as a large sink of carbon that can mitigate climate change. Soil organic carbon sequestration is a proven, economically viable natural climate change mitigation strategy with multiple co-benefits. However, far less attention has been placed on soil inorganic carbon. The top meter of world soils holds around 2200 Pg soil carbon, the largest terrestrial store of carbon, of which approximately 1500 Pg is soil organic carbon. Its counterpart, soil inorganic carbon, has been estimated at somewhere from 700 to more than 1000 Pg (Batjes, 2014; Eswaran et al., 1995). The amount of SIC and SOC stocks down to 2m of soils is comparable (Zamanian et al., 2021) and SIC has a relatively long mean residence time under natural conditions making it highly relevant in the discussion of soil and climate change (Kuzyakov et al., 2019; Zamanian and Kuzyakov, 2019).

About 30–40% of the world's soils are in arid, semi-arid, and hyper-arid areas (Gaur and Squires, 2018; White and Nackoney, 2003). In these areas, SIC can be as important as organic carbon as a carbon store contributing to the CO<sub>2</sub> emissions, as well as, influencing soil physical and chemical properties (e.g., aggregate stability and porosity, water infiltration, pH buffering capacity, and electrical conductivity). SIC comprises up to 90% of soil C pool in these arid areas and in some cases, can contribute up to about 20% to 60% of the whole CO<sub>2</sub> emission from topsoil and subsoil, respectively (e.g. Cardinael et al., 2020). While the rate of CO<sub>2</sub> forming SIC needs further research, notable inorganic carbon sequestration has also been reported (Bughio et al., 2017; Fa et al., 2016; Monger et al., 2015; Okyay and Rodrigues, 2015).

Under natural conditions, it is expected that the SIC mean residence time (MRT) is millennial. However, increased anthropogenic influences could alter the MRT of SIC to annual and decadal time scales (Kim et al., 2020). Recent studies have reported a remarkable contribution of SIC efflux to atmospheric CO<sub>2</sub> due to human activities, including agricultural practices and fertilization (Kim et al., 2020; Ramnarine et al., 2012; Zamanian and Kuzyakov, 2019; Zamanian et al., 2018). Despite its importance, research on SIC is much less than SOC. A search of the Web of Science bibliometric database shows that from 288,632 papers published with the topic "soil" from 2017 to March 2022, 19% of the papers (53,866 papers) focused on carbon. Of those papers on soil carbon, only 8% considered inorganic carbon or carbonate.

The spatial (vertical and horizontal) variations of SIC at local to global scales are also poorly studied. For example, Lamichhane et al. (2019) reviewed 120 papers on digital mapping of soil carbon and reported that only a few studies have considered SIC and only one study was specifically

related to SIC mapping. This lack of attention to SIC is further complicated by the dominance of available models for spatially quantifying SOC that are inadequate for SIC. Both temporal and spatial variations of SIC can be quite different from SOC, due to horizontally and vertically skewed geographical distribution of SIC because of its pedogenic and geologic origins.

While SOC has been featured prominently in global climate change discussions, SIC has not generally been recognized as an indicator or mitigator of change. Most countries do not have detailed maps of SIC and lack assessments showing if land management systems are operating sustainably in relation to SIC and how they will respond to climate change. Agricultural practices such as irrigation can dissolve and remobilize SIC, while application of fertilizers along with lime can change the soil pH and affect the equilibrium between precipitated and dissolved  $Ca^{2+}$  and  $Mg^{2+}$  and their interactions with  $CO_2$  in soil solution via  $HCO_3^-$ . Therefore, accurately modeling the fate and balance of SOC and SIC will allow us to predict how future soil and environmental change will affect the spatial distribution of soil carbon and the functioning of soils. Given the economic and environmental significance of soil and the reliance on soil carbon functions, comprehensive and detailed soil carbon maps (both organic and inorganic) are urgently needed.

Thus, this review aims to:

- (1) Outline the contribution of SIC in the global C cycle.
- (2) Provide a survey of SIC distribution and mapping efforts around the world.
- (3) Provide a comprehensive review of digital mapping and modeling of SIC to identify the current state of the art, research trends and knowledge gaps.
- (4) Discuss factors influencing spatial and vertical distribution of SIC, clarify SIC and SOC relationships, and discuss the effects of climate change on the SIC pool.
- (5) Identify research needs for furthering our understanding of the spatiotemporal distribution of SIC.

# 2. Soil inorganic carbon forms

Understanding the natural and anthropogenic processes driving SIC dissolution or re-precipitation is important for calculating soil C budgets,

predicting spatiotemporal changes in the total soil C pool, and modeling future scenarios of the global C cycle. Generally, SIC is classified into three groups, based on its origin and formation processes.

Geogenic or lithogenic carbonates: Carbonates inherited from calcareous parent materials such as limestone and marls. Geogenic carbonates exist as detrital fragments physically derived calcareous parent materials in place (*in situ*) or as detrital particles transported from other locations (*ex situ*) such as calcareous dust or landslides.

*Biogenic carbonates*: Carbonates formed within terrestrial animals and plants as part of their skeleton or body parts, for example, shells, calcified seeds, root hairs, fungal hyphae, bacteria, or released from or within certain organs such as the esophageal glands of earthworms.

*Pedogenic carbonates*: Carbonates formed authigenically (formed in place from weathering processes) in soils. Pedogenic carbonates form via two pathways: (1) the dissolution and re-precipitation of the SIC pool (i.e., geogenic, biogenic or previously formed pedogenic carbonates, such as nodules) and (2) by the precipitation of  $Ca^{2+}$  from silicate weathering and bicarbonate (HCO<sub>3</sub><sup>-</sup>) from root and microbial respiration in alkaline soils (Fig. 2). Pedogenic carbonate exists in multiple forms and morphogenetic stages that include filaments, coatings, nodules, root casts, pendants, masses, concretions, calcic horizons, and petrocalcic horizons containing laminae, ooliths, pisoliths, and breccia (Ditzler et al., 2017; Gile et al., 1966). These are called caliche, travertine, kankar, or calcrete (Wilford et al., 2015). The carbon contained in all these types is collectively referred to as "soil inorganic carbon." Fig. 1 shows a diagram of SIC groups from a broad viewpoint.

The pedogenesis of SIC and the concept of "carbonate generations" can help understand the processes affecting its role as a source and/or sink of  $CO_2$  (Monger et al., 2015). The forms and formation processes of carbonates with specific emphasis on pedogenic carbonates are discussed in Zamanian et al. (2016).

Pedogenic carbonate is derived from the precipitation of carbonate ions derived from the root and microbial respiration reacting with calcium and magnesium from the soil. In one case, the following reactions show the general equilibrium processes:

$$H_2O + CO_2 \leftrightarrow H_2CO_3 \leftrightarrow HCO_3^- + H^+$$
 (1)

$$CaCO_3 + H_2O + CO_2 \leftrightarrow 2HCO_3^- + Ca^{2+}$$
(2)





**Fig. 1** Broad SIC grouping. Each type enters the soil by various in situ and ex situ pathways. Biogenic carbonate can occur as both lithogenic (e.g., marine shell fragments) and pedogenic (e.g., calcified fungal hyphae) categories. The carbon itself that resides in all types shown above is collectively referred to as soil inorganic carbon. Routine lab analysis involving acid dissolution does not distinguish and separate the various carbonate types. *Figure modified from Monger, H.C., Kraimer, R.A., Khresat, S., Cole, D.R., Wang, X., Wang, J., 2015. Sequestration of inorganic carbon in soil and groundwater. Geology 43, 375–378.* 

The reactions are controlled by carbon dioxide (CO<sub>2</sub>), pH, Ca<sup>2+</sup> content and water content. An increase in soil CO<sub>2</sub> or a decrease in soil pH would lead to the dissolution of carbonate, whereas an alkaline environment would favor the formation of carbonate ions. In addition, the availability of soluble Ca<sup>2+</sup>/Mg<sup>2+</sup> is required for the precipitation of Ca/Mg carbonates. If the cations are from the dissolution of pre-existing carbonates, then reactions (1) and (2) apply. If the cations are from the chemical weathering of silicates, then the unidirectional reactions (3) and (4) apply (Fig. 2):

$$2\mathrm{CO}_2 + 3\mathrm{H}_2\mathrm{O} + \mathrm{CaSiO}_3 \rightarrow \mathrm{Ca}^{2+} + 2\mathrm{HCO}_3^- + \mathrm{H}_4\mathrm{SiO}_4 \qquad (3)$$

$$Ca^{2+} + 2HCO_3^- \rightarrow CaCO_3 + CO_2 + H_2O \tag{4}$$



**Fig. 2** Comparison of the two major pathways of pedogenic carbonate formation. The unidirectional Ebelmen-Urey Reaction involving the weathering of silicates vs the equilibrium reaction involving the dissolution and re-precipitation of existing carbonates. *From Monger, H.C., Kraimer, R.A., Khresat, S., Cole, D.R., Wang, X., Wang, J., 2015. Sequestration of inorganic carbon in soil and groundwater. Geology 43, 375–378.* 

## 3. SIC controlling factors

SIC acts as a terrestrial C store and influences soil properties, and thus, we need to understand its spatial and temporal controlling factors before we can work out its spatial distribution (Table 1). Fig. 3 presents a summary of the current understanding of SIC controlling factors as a function of spatial scale. Pedogenic carbonates mainly accumulate in semiarid or arid soils where evapotranspiration is greater than precipitation. Thus climatic factors of mean annual temperature (MAT), mean annual precipitation (MAP), and mean annual evapotranspiration are important indicators on both regional and global scales (Raheb et al., 2017; Zamanian et al., 2016).

As a climate-driven interpretation and distinction of carbonatic soils, classical concepts pedocal and pedalfer soil zones were developed in the United States (Marbut, 1935). Pedalfers refer to soils in the humid regions enriched with Fe and Al sesquioxides; and pedocals refer to soils where

# Table 1 Summary of studies on controlling factors of SIC. Controlling factors (the ones in

Country	Climate	Controlling factors (the ones in bold were found as the most important)	IC and OC relationship	Method of assessment	Max soil depth (cm)	References
China	Semi-arid continental Afforestation NA Statistical comparison		100	Chang et al. (2012)		
Australia	Sampling location: humid       Liming and wheat plant effects       NA       Analysis of variance, time 2         subtropical with a semi-arid       on DIC leaching and C-efflux       effect, Tukey's HSD and         influence [in Wagga Wagga,       Student t-test		25	Ahmad et al. (2013)		
China	Arid and semi-arid continental monsoon	Land use and soil type	NA	Comparison of SIC densities in different sites	100	Tan et al. (2014)
China	Temperate semi-arid continental monsoon	Desertification stages	Positive	Analysis of variance, linear regression	30	An et al. (2019a)
Meta-analysis of 29 studies	Arid and semi-arid regions	<b>Land use change</b> & indirect factors: pH, MAT, MAP, SOC, soil sampling depth	In some cases positive	Response ration & correlation	100	An et al. (2019b)
Spain	Typical Mediterranean continental	Land use (cropland & rangeland), soil redistribution	In rangeland: negative In cropland: positive (neither was significant)	Pearson correlation, least squares of the general linear models	55	Gaspar et al. (2019)
Iran	Semi-arid	Soil basal respiration, dehydrogenase enzyme activity, bacteria population + physicochemical properties group	Negative	Pearson correlation, Variation partitioning analysis	15	Sharififar et al. (2019a)

China	Semi-arid	Erosion (sediment transport and its influencing factors: raindrop size, <b>PSD<sup>3</sup></b> , <b>runoff rate</b> , inflow rate)	Negative	Boosted regression tree & Pearson correlation	20	Wang et al. (2019)
China	Subtropical monsoon	Reclamation duration, land use change, physio-chemical properties such as CaO, Fe & Al oxides, C:N, texture	NA	Random forest model	20	Zhang et al. (2019)
China	Arid to semi-arid	-arid MAT, MAP, geographic Negative in all Structural 1 position (latitude, longitude, depth layers equation modeling altitude), soil type, pH, <b>moisture</b> , land use, climate, <b>OC</b>		100	Zhao et al. (2019)	
Cypress	Subtropical Mediterranean and semi-arid	<b>Lithology</b> , land use/cover, major and trace elements, SOC, climatic factors	NA <sup>2</sup>	Spatial correlation using variography	20	Zissimos et al. (2019)
China	Cold semi-arid	MAT, <b>MAP</b> , <b>pH</b> , total N, total C, C:N, soil water content, DON, TDN, SOC:SIC, NH <sup>4</sup> <sub>4</sub> -N, NO <sub>3</sub> -N	Negative	Generalized linear model	70	Du and Gao (2020)
Spain	Sub-humid Mediterranean mountain agrosystem	Runoff/water erosion: land use & vegetation cover; curvature & flow accumulation: all important	Na	Analysis of variance, <sup>137</sup> Cs inventories, perceptual modeling, Kruskal-Wallis test	5	Gaspar et al. (2020)
China	Temperate Semi-arid continental monsoon	Soil erosion types, SOC	At erosional sites: negative At depositional sites: positive	Partial least squares regression & analysis of variance	600	Tong et al. (2020)

Country	Climate	bold were found as the most important)	IC and OC relationship	Method of assessment	Max soil depth (cm)	References
Meta-analysis	Polar, temperate, sub-tropical, tropical	MAT, MAP, <b>land use</b> , <b>basin</b> <b>surface area, morphology of</b> <b>the river basins</b> : on dissolved IC transitioning by water	NA	Mata-analysis; correlation	_	Chaplot and Mutema (2021)
Italy	Semi-arid Mediterranean	Mediterranean       Factors affecting IC       NA       Kinetic and         accumulation and       thermodynamic         CO2 sequestration: Relief,       geochemical model, ion         distance from gypsum outcrops,       activity, log Q/K ratio,         parent material, Ca <sup>2+</sup> from       Gibbs free energies         gypsum       Gibbs free energies		150	Laudicina et al. (2021)	
China	Semi-arid continental monsoon	Revegetation: enhancing <b>N</b> , available K, CEC, <b>clay</b> , <b>Ca<sup>2+</sup>,</b> <b>Mg<sup>2+</sup></b>	Positive	Analysis of variance, linear regression, Redundancy analysis	20	Li et al. (2021)
South Africa	Arid and semi-arid	Grazing intensity, soil particle size distribution	NA	Analysis of variance, Tukey's HSD test	5	Loke et al. (2021)
Iran	Semi-arid	Elevation, slope, stream power index, <b>topographic wetness</b> <b>index, alkaline cations</b>	Negative	Analysis of variance, stepwise multiple regression, Pearson correlation	20	Mohseni and Salar (2021)
China	Subtropical to warm temperate monsoon	<b>Particle size</b> , pH, salinity, <b>N</b> , P, S	Positive	Partial least squares regression	3	Yang et al. (2021a, 2021b)

 Table 1 Summary of studies on controlling factors of SIC.—cont'd

 Controlling factors (the ones in

Note: The review of "SIC controlling factors" is not a comprehensive survey, but important examples from different countries.



Fig. 3 Summary of SIC controlling factors in spatial scales as to inherent soil characteristics, anthropogenic factors and geographical variables.

evapotranspiration is higher than precipitation with accumulation of carbonates. The continental USA was divided into west dry zone and eastern humid zone based on the pedocal–pedalfer soil boundary called the Marbut line, where mean annual precipitation and evapotranspiration were equal. The Marbut line has been revised using better climatic data and defined in terms of annual water balance equal to zero (D'Avello et al., 2019; Salley et al., 2016). D'Avello et al. (2019) found that the annual water balance was positively related to the depth of maximum CaCO<sub>3</sub> accumulation and negatively related to SOC stock. However, this moisture relationship was weak as other factors also control SIC and SOC distribution, such as loess deposition and palæoclimatic environments.

Low rainfall, high temperature and evaporation lead to the retention of atmospherically sourced calcium. However, climate is not the sole controller of carbonate distribution. Proximity to the coast and parent materials are also important factors controlling the distribution of SIC. The type of soil minerals can determine the source of Ca, which can influence SIC formation. Additionally, vegetation could be an indicator of SIC, as soil  $CO_2$  is associated with the respiration of plant roots, decaying organic matter and microbial activity. Microorganisms could also have a role in SIC accumulation by acting as catalysts for soil carbonate dissolution.

Mangrove forests on coastal wetlands, which are known for accumulating a high organic carbon, also have a reasonable amount of biogenic inorganic carbon derived from sediments of calcifying organisms (Alongi, 2014; Ouyang and Lee, 2020). SIC controlling factors vary at different spatial extents, although some overlapping might occur (Fig. 3). For example, at micro to *meso* scales, Ca<sup>2+</sup> availability, pH (in rhizosphere or bulk soil), clay and water content are crucial. At the soil profile to field scale, pH, soil depth, precipitation, leaching, respiration, microbial activity, soil type, nutrients, acid rain, salinity, land use, and fertilization are potential factors controlling SIC (Ferdush and Paul, 2021). At the landscape scale; topography, land cover, precipitation and temperature are important factors (Guo et al., 2006b). For example, land cover (forest, rangeland, cropland, etc.) with regard to soil management (irrigation, fertilization and tillage) and inherent soil pH can influence carbonate dissolution/precipitation. Topography contributes to deposition and leaching of surface carbonates. At the macro scale (regional to continental scales) climate, land use and lithology are the main drivers of variations in SIC stocks (Fig. 3).

Irrigation as an anthropogenic factor can be an influential factor on SIC. It improves biomass production and enhances biological activity, i.e., via root and microorganism respiration and increasing  $CO_2$  partial pressure. In turn, it may either lead to enhanced trapping of  $CO_2$  in carbonate-free soils through precipitation of pedogenic carbonates following increased silicate weathering and release of  $Ca^{2+}$  ions or decreasing soil pH in carbonate-containing soils, which increases the dissolution of carbonates and  $CO_2$  emission (Gocke et al., 2011). Nitrogen fertilization and liming can affect SIC stocks dynamics, which are discussed in the following sections.

A somewhat ignored factor in assessing SIC spatiotemporal variation is SOC content (Guo et al., 2016; Wang et al., 2015; Yang et al., 2021a,b; Zhang et al., 2010). SIC and SOC can have positive and negative correlations in different climates and geographic regions. SIC–SOC relationship is discussed in detail, further in this review.

# 4. Vertical distribution of soil inorganic carbon

Unlike SOC which accumulates in the topsoil, the precipitation of SIC occurs mainly in the deeper soil profile. Different trends of depth variation for SIC have been reported in the literature. Land use, soil type, precipitation regimes and erosion/deposition (soil redistribution) all affect the vertical distribution of SIC. The depth distributions of SIC and SOC are usually opposite (Fig. 4). An exponential decrease of SOC and a linear increase for SIC with depth has been reported for Belgium with about



**Fig. 4** An example of depth variation in a Calcarosol in Australia (Edgeroi area, New South Wales); blue line depicts carbon as CaCO<sub>3</sub> and red line depicts carbon contained in SOC (the black horizontal lines show standard errors of measurements at several depth layers).

13,000 soil profiles, where the main controlling factor for SIC variation was geology (Lettens et al., 2004). Batjes (2004) reported an increasing SIC content with depth at regional scales such as arid zones of Kenya as well as in other parts of the world (Batjes, 2014). At the landscape scale, both increase and decrease of SIC content toward the subsoil ( $\approx$ 30–60 cm depth) have been reported (Liu et al., 2014; Wang et al., 2010; You et al., 2020). The changing trend below 60 cm, however, was mostly related to the soil type (Jiménez-Aguirre et al., 2018; Sharififar et al., 2019a). Changes in the amount of SIC in subsoil layers are more conspicuous in arid and semi-arid regions (Raheb et al., 2017; Sharififar et al., 2019a), where low rainfall precipitation leads to carbonate accumulation in the subsoil.

The depth of pedogenic carbonate in a soil profile is commonly related to rainfall (Retallack, 1994). Jenny and Leonard (1934) found that depth to carbonate (DTC) is a function of mean annual precipitation (MAP), where DTC increases from 350 to 1000 mm of MAP on Mollisols in the central Great Plains of the United States. Retallack (1994) derived a MAP-DTC relationship based on observations of different soil types. This relationship



**Fig. 5** Depth to carbonate (DTC) layer as a function of Mean Annual Precipitation (MAP) using four regression equations from Retallack (1994; equation a), Arkley (1963; equation b), Royer (1999; equation c), Jenny and Leonard (1934; equation d). The graphs are schematic representations of approximate carbonate horizon depth as related to mean annual precipitation. Note that the graphs are redrawn based on findings from the literature from the United States.

was further improved by Royer (1999) with a larger dataset which found that the depth to the carbonate horizon increased by 360mm with every 100mm increase in precipitation (Fig. 5).

Accounting for SIC stock is essential for inventory and calculation of carbon fluxes. SIC storage is calculated for a specific depth layer as follows:

$$SIC_{stock(i)} = SIC_i \times BD_i \times T_i \times (1 - CF_i)$$
(5)

where, SIC<sub>stock(i)</sub> is SIC stock in layer *i* (kg m<sup>-2</sup>), SIC<sub>i</sub> is the gravimetric SIC content (kg kg<sup>-1</sup>) in <2 mm fraction,  $BD_i$  is the bulk density of the whole soil of layer *i* (kg m<sup>-3</sup>),  $T_i$  is the thickness of the layer (m) and  $CF_i$  is the coarse fragment content (>2 mm) of the layer (kg kg<sup>-1</sup>).

If CaCO<sub>3</sub> is considered an indicator of SIC, the storage of SIC can be calculated as:

$$SIC_{stock} = (CaCO_3 \times 0.12) \times BD_i \times T_i \times (1 - CF)$$
(6)

where  $CaCO_3$  is the calcium carbonate equivalent (CCE) concentration (kgkg<sup>-1</sup>), 0.12 is the conversion factor for CCE to SIC. The SIC stock can be computed for specific areas or biomes by multiplying  $SIC_{stock(i)}$  by the area (Rasmussen, 2006).

SIC stock is usually calculated for fractions <2 mm, however, inorganic C can exist as coarse fragments, which are usually considered as gravel or rock. Thus, the actual IC stock in the soil profile could be much greater than that being measured. More detailed volumetric sampling and measurement are required to remedy this problem.

# 5. Global distribution of SIC and examples from different countries

We need precise and accurate maps of SIC for baseline soil carbon specifications, distribution, modeling, and sequestration/emission assessments for agricultural management and climate mitigation. A global interpretative map of SIC stocks in the top 1 m of soils provided by the United States Department of Agriculture (Eswaran, Van Den Berg & Reich, 1993) (Fig. 6) shows that SIC amount is comparatively larger in areas



**Fig. 6** Global variation of SIC to the depth of 1 m (from United States Department of Agriculture; http://nrcs.usda.gov). Map uses the Equal Earth projection.



Fig. 7 Average SIC and SOC stocks as a function of latitude. Data from global SIC and SOC maps of the United States Department of Agriculture (https://www.nrcs.usda.gov).

with low precipitation. Fig. 7 shows the variation in SIC and SOC stocks as a function of geographical latitude. Largely in the northern latitudes and down to  $40^{\circ}$  south ( $-40^{\circ}$ ), an overall inverse relationship can be observed between SIC and SOC stocks. These variations show that SIC stocks are high, where SOC stocks are low and vice versa. This signifies the role of precipitation/ evapotranspiration and climate zones on soil C dynamics.

Globally, the distribution of calcareous or carbonate-containing soils is still uncertain with estimates varying from 13% to 40% of the Earth's land surface (Gaur and Squires, 2018; White and Nackoney, 2003; Wilford et al., 2015). This large uncertainty is due to inaccurate mapping representations and differences in definitions. Currently, there is no effort in global data collection of SIC information and no endeavour to improve global SIC mapping with a fine granular resolution. The map in Fig. 6 was created more than three decades ago based on interpretive global soil type maps and climate (Eswaran et al., 1993). There is a lack of knowl-edge on the current stock, and probable loss or change in SIC. Thus, the following sections provide some details about SIC mapping efforts from different countries with significantly vast areas of soils containing SIC. While this is not an exhaustive list, the survey aims to give a snapshot of the status of SIC mapping in different countries. The case studies were arranged from South to North.

#### 5.1 SIC in Australia

In Australia, SIC is predominantly located in arid and semi-arid regions due to low rainfall and biological activity (Ahmad et al., 2015; Wilford et al., 2015). The lack of rainfall results in carbonate relocation in the soil profile rather than being lost due to leaching and is coincident with the lack of biological activity reducing acidity inputs. The retention of Ca<sup>2+</sup> and Mg<sup>2+</sup> ions in these areas increases the probability for SIC precipitation and persistence. In addition to in situ geological formations, the sources of Ca<sup>2+</sup> and Mg<sup>2+</sup> ions include rainfall, sea-spray and eolian dust. About 90% of the calcium in Australian soil carbonate is attributed to marine sources (Wilford et al., 2015). The persistence of SIC to the south of 30°S is sourced from winter dominated rainfall sourced from rain washed carbonates over the Indian and Southern Oceans, which have significant levels of  $Ca^{2+}$  and  $Mg^{2+}$  (Wilford et al., 2015). The high rainfall in regions of the eastern margin of Australia, southwest Western Australia, Tasmania and Northern Tropics results in SIC being rare or absent compared to the rest of the continent (Fig. 8).

Soils with carbonate have been estimated to cover around 21–50% of Australia land surface. A digital soil mapping (DSM) approach in Australia showed different regions of distinct CaCO<sub>3</sub> content that resulted in a combination of climate, soil type, parent materials, and vegetation (Wilford et al., 2015). Some distinct areas include eolian-derived carbonate in western New South Wales (NSW), regolith carbonate in a pocket of the Eastern Highlands in southeast Australia, known as the Monaro Plains, Ca-rich Vertosols in arid and semi-arid regions of NSW and Queensland (QLD). Areas with the largest SIC is in the shallow, highly calcareous soils



**Fig. 8** Soil calcium carbonate map of Australia. The yellow line represents the Menzies line delineating a major change in vegetation, and soil and hydrology. *Based on Wilford, J., De Caritat, P., Bui, E., 2015. Modelling the abundance of soil calcium carbonate across Australia using geochemical survey data and environmental predictors. Geoderma 259, 81–92.* 

over the Nullarbor Plain in the south of Western Australia dominated by calcrete, where karstic landforms dominate this region. Other highly calcareous soils are in the Flinders-Mount Lofty ranges and Eyre Peninsula in South Australia.

In the semi-arid region, the soil types also affect the presence and persistence of SOC and SIC. As an example of the profile variations in different soil types, Fig. 9 shows the SIC and SOC profile variations in the Edgeroi area in New South Wales, with 9 soil orders according to the Australian soil classification system (Isbell, 2016). Calcarosols (equivalent to Aridisols and Alfisols in the USDA soil taxonomy) show the highest variation and the largest amount of SIC. The changing trend in SIC content in Chromosol (equivalent to Alfisols and some Aridisols), Vertosols (equivalent



Fig. 9 Example of depth variation of soil IC and OC in Australian soils. Each of the displayed soil profile is an average data from several soil profiles.

to Vertisols), Dermosols (equivalent to Alfisols and Ultisols) and Sodosols (equivalent to Alfisols and Aridisols) was increasing down to 60 cm depth and then decreasing down to 100 cm depth. Kurosols (equivalent to Ultisols and Alfisols) and Tenosols (equivalent to Inceptisols, Arisidols, or Entisols) show the lowest accumulation of SIC with no obvious depth-related changes (Fig. 9). The trend of SOC profiles generally shows an inverse relationship with SIC amounts, decreasing down to 20 cm depth.

Climate change scenarios indicate that Australian soil will experience drying by 7% between 2020 and 2050. About 6.5 million km<sup>2</sup> (86% of the land) are estimated to suffer from soil moisture decline of about 10% from 2030 to 2040 as the driest decade (Guglielmo et al., 2021). This will have an impact on SIC distribution.

#### 5.2 SIC in Chile

Soil carbonates are common in the dry region of South America, especially in Chile and Argentina. According to the literature, in Chile, soils rich in pedogenic carbonates occur between 18°S and 34°S (Díaz Vial et al., 1960; Pfeiffer et al., 2011), and they reappear in the arid sections of the southern Patagonian Steppe around 52°–53°S (Bockheim and Douglass, 2006; Filipová et al., 2010; Vial, 1965). This distribution is highly correlated with Chile's climatic gradient. The climatic gradient in Chile is extreme and follows a latitudinal gradient from north to south, ranging from the hyperarid environment at the Atacama Desert to the very humid conditions of Patagonian Fjords.

In the Atacama desert, precipitation is  $<20 \,\mathrm{mm}$  of mean annual precipitation (MAP) below 2000m above sea level and occurs during sporadic events such as once per decade. Above this altitude, precipitation increases accordingly, rarely exceeding 100 mm/y with rainfall mostly falling during summer, influenced by the south American Monsoon (Garreaud et al., 2009). One of the characteristic features here is the near absence of plants and the presence of highly soluble salts in the soil profiles as a result of accumulation throughout millions of years of atmospheric deposition (Finstad et al., 2014). These soils contain very low levels of calcium carbonates, as Ca is sequestered into gypsum and anhydrite due to an abundance of  $SO_4$  and the very low levels of  $CO_2$  in soils (Ewing et al., 2006, 2008; Quade et al., 2007). It has been proposed that most of the carbonates are a relict feature from wetter climates with a minimum plant coverage on the surface (Ewing et al., 2006). However, there is evidence of modern occurrences based on  $\delta$ C13 isotopes with low levels of calcium carbonates in which the carbonatic source comes most probably from atmospheric and microbial respiration (Finstad et al., 2016; Pfeiffer et al., 2019). Based on Sr isotope data, the calcium source of carbonates in these soils mostly comes from atmospheric deposition (Rech et al., 2003), as it does not exist in the desert (Owen et al., 2011).

Toward the southern border of the desert, near the modern section of 20–50 mm MAP, there is a transition from gypsic to calcic soils from north to south (drier to wetter). Here, geochemical data indicate that carbonates formed mostly during wetter periods during the Pleistocene (Pfeiffer et al., 2021). Azonal soils in the Atacama Desert influenced by groundwater recharge tend to form carbonate accumulations due to upward flow during evaporation (Finstad et al., 2016; Pfeiffer et al., 2018).

South of 26°S, where the Mediterranean biome is predominant, calcium carbonate soils are a ubiquitous feature in the landscape. This area covered the entire range of precipitations where carbonates are formed according to the literature (Retallack, 1994; Royer, 1999), while between 26°S and 33°S winter rainfall occurs sporadically, creating a blooming desert, which might contribute as the main source of  $CO_2$  since the early Pleistocene (Ebeling et al., 2016). South of 33°S is an area of marked seasonal precipitation typical of Mediterranean climates with dry summers and wet winters, here carbonatic soils develop on a vast array of parent materials and geomorphic positions, such as andesitic and granitic Hillslopes (Vera, 1985), marine

terraces on calcareous rocks (Pfeiffer et al., 2011, 2012), alluvial deposits (Aburto et al., 2008), andesitic hillslopes (CIREN, 1996a) and palustrine environments (CIREN, 1996b; Vargas De la Maza, 2022). The southernmost soil with the presence of calcium carbonate corresponds to a lacustrine origin soil with a MAP of 790 mm (CIREN, 1997). South of this location, precipitation increase systematically toward the south, consequently disappearing the climatic conditions that allow pedogenic carbonates to form. This incompatibility for calcium carbonates to form is increased by the dominance in the landscape of Andisols derived from volcanic ash, which tends to be acidic.

In Chilean Patagonia, a common characteristic is a complete time "reset" after the last glacial maximum (LGM), as the entire region was covered with ice then. The region's extent and intensity of glacial erosion resulted in the absence of soils older than the LGM. Over this young landscape acts a climatic gradient with precipitation that ranges from 8000 mm in the western fjords to 200 mm in the eastern pampas allowing very different soils to converge in a few kilometers. A particular feature of carbonatic soils in this region is that they are formed over glacial and fluvioglacial deposits (Bockheim and Douglass, 2006; Douglass and Bockheim, 2006) and many of them present cryogenic properties (Bockheim et al., 2009). While precipitation could be lower enough to form well-developed calcic soils, they show early development stages (Bockheim and Douglass, 2006; Filipová et al., 2010), most probably related to their young age and the low temper-atures predominant in the area.

A recent digital soil mapping estimate of carbon stocks for Chile up to 1 m depth (Fig. 10), estimated SIC by around 1.4 Pg, being only a small fraction of SOC stock of Chile which was around 21.5 Pg (Padarian et al., 2022). Given the observed increasing content of SIC in depth, we expect that the stock would increase considerably at lower depths. This highlights the need for specific surveys with observations beyond the traditional 1 m depth.

Intensive agriculture may accelerate the loss of SIC. For example, Kim et al. (2020) found a decrease of 36% in SIC stock at a Pampas grassland site at 34°S in Argentina. Future climate change scenarios suggest that there would be a small decrease or no change in rainfall over Chile but a clear warming trend over northern and inland regions (Williams, 2017). It is still unclear how these will affect SIC stock in the region.



Fig. 10 Distribution of SIC stock to 1 m depth in Chile.

## 5.3 SIC in South Africa

Calcareous soils in South Africa are quite widespread. In the local classification system, calcareous soils are placed in 31 out of the 145 recognized soil types (Soil Classification Working Group, 2018). The climatic border of the calcareous soils is regarded as the area with a Weinert N-value >5 (Fig. 11; Weinert, 1980). The Weinert N-value is calculated as in Eq. (7):

$$N = 12 \times Ej/MAP$$
(7)



**Fig. 11** Weinert N-values for southern Africa. *Based on Jordaan, G.J., Kilian, A., Muthivelli, N., Dlamini, D., 2017. Practical application of Nano-technology in roads in southern Africa, in: Proceedings of the 8th Transportation Technology Transfer (T2) Conference, Lusaka, Zambia. pp. 8–10.* 

where N is the Weinert N-value, Ej is the evaporation in January, the hottest month in South Africa, and MAP is the mean annual precipitation.

An interesting phenomenon that leads to calcareous soils is the occurrence of giant earth mounds locally called heuweltjies (Afrikaans for small hills), created by the southern harvester termite *Microhodotermes viatorare* (Picker et al., 2007). These heuweltjies occupy 14–25% of the land in the southwestern parts of south Africa (Lovegrove and Siegfried, 1986). The soils within these heuweltjies are often calcareous, due to the Ca<sup>2+</sup> and  $CO_3^{2-}$  enrichment by termite activity, as well as the decrease in soil moisture due to the mound structure (Fey, 2010). The increased fertility of the heuweltjies and the drier soil moisture regime give rise to a different vegetation structure, clearly visible on satellite images (Fig. 12).

Anecdotal evidence suggests that calcareous soils also occur within the first dune from the sea, all along the South African coast, from the area west coast to the tropical east coast, due to carbonates being wind-blown from the sea (F. Ellis, personal communication). The only exception is in the southern and eastern cape, where the process of acidification through podzolization overrides that of calcification.



Fig. 12 Typical heuweltjie patterns from a Google Earth image near Robertson in the Western Cape of South Africa.

There have been efforts to map the calcareous soils in South Africa, and a distribution map could be created from the land type survey (Fig. 13; Land Type Survey Staff, 1972–2002), which was conducted to determine the agricultural potential of South Africa's soils. The map units are polygons grouping areas with similar macroclimate, geology and soil distribution patterns at a scale of 1: 250,000 (Paterson et al., 2015). Each polygon has an accompanying inventory showing an approximation of the percentage cover of different soil types on each terrain unit within the land type. Therefore, the final map gives an idea of the distribution of calcareous soils, rather than an accurate depiction of exactly where they occur. No attempt has been made to quantify the carbon stocks, although it could be vast as the soils cover large areas and the thickness of the calcareous layers can vary between a few millimeters up to 200 m thick.

## 5.4 SIC in the Mediterranean basin

In the Mediterranean basin, soils are mostly developed on calcareous rocks. Calcisols, Leptosols, Cambisols and Luvisols (Food and Organization of the United Nations, 2014) are dominant soil types (Jones et al., 2005) and often contain various morphologies and concentrations of carbonates. Areas of Mediterranean conditions cover 5% of the world's SIC stock. The major types of carbonate minerals are calcite (CaCO<sub>3</sub>) and dolomite (MgCO<sub>3</sub>). For example, Díaz-Hernández et al. (2003) measured



**Fig. 13** Distribution of the percentage of carbonatic soils in South Africa at a national scale based on interpreting polygons of the land type survey (Land Type Survey Staff, 1972–2002; Le Roux et al., 2013).

81 profiles located in Southeast Spain and reported an average of  $134 \text{ kgm}^{-2}$  (1340 t SIC ha<sup>-1</sup>), of which  $88 \text{ kgm}^{-2}$  was in the form of calcite and  $46 \text{ kgm}^{-2}$  in the form of dolomite in the 0–2 m depth layer.

In the Near East North Africa (NENA) region, calcareous soils are common and the stocks of SIC are often higher than those of SOC. In the topsoil (0-0.3 m), the SIC stock ranged between 25 and  $120 \text{ tha}^{-1}$  compared to  $12-35 \text{ tha}^{-1}$  for SOC. In subsoils (0.3-1 m), the difference in stocks is even larger, the SIC ranged between 2.5 and  $45 \text{ kgm}^{-2}$ , compared to  $2-4.5 \text{ kgm}^{-2}$  for SOC (Darwish et al., 2018). Under Mediterranean conditions, the SIC stocks show high spatial variation in different geographic locations. For example, Lebanon, Gaza, Jordan, Mauritania, Sudan show relatively low SIC stocks with average values lower than  $5 \text{ kgm}^{-2}$  in topsoil and lower than  $10 \text{ kgm}^{-2}$  in subsoils, while Egypt has much higher values with average mean stocks exceeding  $10 \text{ kgm}^{-2}$  in topsoil and  $20 \text{ kgm}^{-2}$  in the subsoil. Brahim et al. (2012) measured SIC contents in all soil groups of Tunisia with high disparities such as reduced concentration (<1 g SIC kg^{-1} soil, or <0.1%) in Podzoluvisols, medium concentration in Solonchaks

and Gleysols with about  $16 \text{ g SIC } \text{kg}^{-1}$  soil, and high concentration in the other soils with mean SIC concentration up to 24–36 g SIC  $\text{kg}^{-1}$  soil in Luvisols and Lithosols.

Densely populated with a semiarid climate and long history of agriculture, Mediterranean soils face several threats to global changes: soil losses, salinization and SOC depletion (Lagacherie et al., 2018). Particular attention must be paid to the effect of climate changes on the Mediterranean region, where increased temperature, increased frequency of extreme events such as heat waves and decreased and more concentrated rainfall are forecast (IPCC, 2007). Increasing temperature and dry-wet cycles have been largely described to affect SIC dynamics (Lapenis et al., 2008; Tao et al., 2022). A study for Northwest Tunisian calcareous soils indicated a moderate and positive response of soil CO2 emission to temperature. Q10 (the proportional change in respiration with a 10C° increase in temperature) was measured at 1.7 and ranged around  $2.6 \pm 1.2$ , given in the literature (Hamdi et al., 2013). At first sight, these Tunisian calcareous soils seem to have no specific response to temperature increases, but extra  $CO_2$  is emitted from both SOC and SIC pools with increasing temperature (Chevallier et al., 2016). There are not enough studies focusing on how and under what conditions the large SIC pool of the Mediterranean soils may affect climate change and vice versa. A better understanding of SIC dynamics and its role in ongoing global changes is particularly critical for Mediterranean areas, where SIC is the most important C form.

#### 5.5 SIC in Iran

In Iran, various climatic situations have created diverse bioregions and soil types (Roozitalab et al., 2018a). In the country's northern provinces, sub-humid and humid climate overshadows the soils and their properties, resulting in lower carbonate accumulation, while in the south, central and eastern areas with arid and semi-arid climate and geological deposition and lithology, there is a much higher inorganic carbon even in topsoil. About 41% of Iranian soils are Aridisols and 41% Entisols (Roozitalab et al., 2018b), which shows that the soils are mostly young in terms of development and that the depth to carbonate accumulation is not as deep as areas with high rainfall. Due to low rainfall in many parts of the country, carbonates leaching has not happened as much as in the northern regions neighboring the Caspian Sea.



**Fig. 14** Example of SIC (IC) vs SOC (OC) variations with depth for different soil types in a typical semi-arid region of Iran. Each of the displayed profile is an average data from several soil profiles. Adapted from Sharififar, A., Sarmadian, F., Alikhani, H., Keshavarzi, A., Asghari, O., Malone, B.P., 2019a. Lateral and vertical variations of soil organic and inorganic carbon content in Aridisols and Entisols of a rangeland. Eurasian Soil Sci. 52. https://doi.org/10.1134/S1064229319090084.

Case studies and technical reports of local research carried out by the National Soil and Water Research Institute show that carbonates are accumulated at around 30 cm depth in a vast majority of soils and often show various increasing and sometimes decreasing trends below this depth. Geomorphology and parent material play a major role in Iranian soil variation and the distribution of SIC (Eghbal et al., 2018). Calcification, and carbonates redistribution in soil profiles are among the major soil development factors in arid and semi-arid areas of the country, which can be the main drivers of SIC spatiotemporal variation (Khormali and Toomanian, 2018). There is an observable trend of SIC variation in different soil types, even in areas with similar climatic and physiographic situations. Fig. 14 shows an example of SIC and SOC depth variations in a typical semi-arid region in Iran with regard to different soil types. As can be seen in Fig. 14, SIC and SOC show inverse and diverse variations with depth. This example shows that SOC usually has a decreasing trend with depth, while SIC can increase or decrease with depth, which is normally expected in arid and semi-arid regions of Iran. Deposition and soil redistribution by surface phenomena, which are perceivable from the soil types, are drivers of such diverse variation in SIC with soil depth (Sharififar et al., 2019a,b). The graphs presented here (Fig. 14) are regarded as typical examples for wide areas of Iran under arid and semi-arid climatic situations and calcareous geological structures/parent material.

As in many other parts of the world, no systematic and sufficient studies have been done on SIC spatiotemporal modeling and mapping in Iran and currently, there is no national digital soil map for SIC. This fact implies the need for further SIC mapping and exploration to understand its interactions with climate change, and C cycle at the national scale.

#### 5.6 SIC in China

Over one-third of China's land is characterized as arid and semiarid biomes (Xia et al., 2017), in which pedogenic inorganic carbonate accumulates. In general, the topsoil SIC contents are positively correlated with aridity and negatively correlated with air temperature (Mi et al., 2008). Therefore, the SIC contents significantly decline from west to east and from north to south of China (Wu et al., 2009). SIC storages in subsoils were much greater than those in topsoils. For the upper 3 m, approximately 42% of the SIC was found in the depth of 2–3 m. The estimated SIC stocks at soil depths of 0–1 m, 0–2 m and 0–3 m were 53.58  $\pm$  0.57 Pg, 134.96  $\pm$  0.91 Pg and 232.21  $\pm$  1.21 Pg, respectively (Song et al., 2022).

Song et al. (2022) conducted spatiotemporal modeling of the SIC pool across China's main ecosystems (i.e., cropland, forest and grassland). Soil samples were collected from 13,769 sites across mainland China in the 1980s (n = 7299), 2000s (n = 774) and 2010s (n = 5696). The paired contrasts were selected from samples in the 1980s and 2010s, with a distance of <50 km, followed by a paired *t*-test to evaluate the changes in SIC by taking the soil samples in the 1980s as the baseline. Overall, the study found a significant decrease in SIC stock in the upper 30 cm of soil surface  $(11.33 \,\mathrm{gCm^{-2} \, year^{-1}})$  during 1980–2010 (Fig. 15A). Over 30 years, the SIC of cropland, forest and grassland declined by  $0.34 \text{ kgm}^{-2}$  (3.4 tha<sup>-1</sup>), 0.38 kgm<sup>-2</sup> and 0.35 kgm<sup>-2</sup>, respectively. Nevertheless, high prediction uncertainty (i.e., high standard deviation values) was found in areas with high SIC (Fig. 15B). For the topsoil (0-30 cm), the mean values of SIC maps in 1980 and 2010 were  $1.64 \pm 0.14 \text{ kgm}^{-2}$  and  $1.50 \pm 0.14 \text{ kgm}^{-2}$ , respectively. During the last three decades, the total SIC pool declined by  $1.37 \pm 0.37$  Pg (8.99  $\pm 2.24\%$ ) (Fig. 15A), most of which may have been released as net  $CO_2$  emission (a maximum of 89.22 Tgyear<sup>-1</sup>).

China has experienced rapid land use changes and intensified agriculture by wide application of chemical fertilizers over the past several decades. Our study (Song et al., 2022) shows that soil acidification caused by global change, atmospheric nitrogen (N) deposition, and N fertilizers has greatly



Fig. 15 (A) SIC changes in topsoil (0–30 cm) across China from 1980 to 2010 (negative values refer to SIC loss); (B) standard deviations of variations in SIC changes. The red line of the histogram at each map indicates the mean value. Adapted from Song, X.-D., Yang, F., Wu, H.-Y., Zhang, J., Li, D.-C., Liu, F., Zhao, Y.-G., Yang, J.-L., Ju, B., Cai, C.-F., 2022. Significant loss of soil inorganic carbon at the continental scale. Natl. Sci. Rev. 9, (nwab120).

accelerated SIC loss. Assuming the spatial controls continue, the potential changes in total SIC stock were projected under two Shared Socioeconomic Pathway (SSP) scenarios, including SSP1–2.6 and SSP3–7.0 of Coupled Model Intercomparison Project Phase 6 (CMIP6) models that represented the low and medium/high radiative forcing, respectively. Overall, approximately 19.12–19.47% of the SIC stock in China, equivalent to 2.69–2.74 Pg C, could be further lost by 2100 when taking the 2010s as baseline. Notably, under SSP1–2.6 scenario, SIC in natural ecosystems (forest and grassland) may decrease by 12.82%, while 31.27% of the SIC in cropland might be lost.

The consumption of SIC may offset a large portion of the efforts of SOC sequestration practices, which emphasizes the importance of an updated SIC inventory for global terrestrial carbon pool estimates. The indirect coupling of reactive nitrogen and inorganic carbon through soil reactions, in addition to routine biotic carbon and nitrogen coupling should be considered for a complete picture of global carbon and nitrogen cycling.

#### 5.7 SIC in France

Using the French systematic soil quality monitoring grid (RMQS, Arrouays et al. (2002)), design-based estimates of the mean and total amounts of SIC content and stocks and their associated uncertainties have been produced following the approach recommended by Brus and Saby (2016). Design-based estimation of means or totals from systematic random sampling is straightforward: the sample mean is an unbiased estimator of the population mean. These estimates are also precise, thanks to the good spatial coverage of the observations. However, this is not the case for the sampling variance of the estimated mean (or total), i.e., the variance of the estimated mean (or total) over repeated systematic random sampling. Therefore, we approximated this quantity by multiplying the simple random sampling approximation by Geary's spatial autocorrelation index (D'Orazio, 2003). This approach is more straightforward to implement than the one retained by Marchant et al. (2015) which required geostatistical simulations.

Here, we provide estimates for 0–0.3 m and 0.3–0.5 m and compare them to SOC estimates (Table 2). We also provide the percentage of sites where SIC>SOC and the mean value of SIC content for these sites. Soil inorganic carbon (SIC) was calculated from the calcimetry analysis (CaCO<sub>3</sub> equivalent) multiplied by 0.12. The detailed analytical protocol is available in Marchant et al. (2015). Fig. 16 shows the distribution of SIC density in France.

Variable	Units	0–30 cm	30–50 cm	0–50 cm
Mean SIC	$\rm g k g^{-1}$	6.55 (0.29)	7.27 (0.35)	6.48 (0.32)
Mean SIC density	$\mathrm{kgm}^{-2}$	1.89 (0.086)	1.421 (0.071)	3.15 (1.48)
Total SIC stock	Pg	1.025 (0.047)	0.712 (0.037)	1.737 (0.082)
Mean SOC	$\mathrm{gkg}^{-1}$	25.18 (0.32)	10.50 (0.22)	17.51 (0.28)
Mean SOC density	$\mathrm{kgm}^{-2}$	6.74 (0.06)	1.98 (0.04)	8.41 (0.09)
Stock_SOC_France	Pg	3.65 (0.034)	0.99 (0.02)	4.64 (0.053)
Area SIC > SOC	%	10.95	13.36	10.5
Mean SIC > SOC	gkg <sup>-1</sup>	44.83(0.94)	41.4(1.06)	43.4(1.01)

**Table 2** Summary statistics of soil inorganic and organic carbon in French soils (including Corsica) obtained from RMQS.

Values represent means and standard errors in parenthesis. The standard errors were corrected with a factor that is a function of Moran's index of spatial autocorrelation. The number of sites with available data used for calculations was 2161 for 0-30 cm, and 1998 for 30-50 cm.



**Fig. 16** Map of topsoil (0–0.3 m) SIC density in mainland France, including Corsica. Based on (Marchant, B.P., Villanneau, E.J., Arrouays, D., Saby, N.P.A., Rawlins, B.G., 2015. Quantifying and mapping topsoil inorganic carbon concentrations and stocks: approaches tested in France. Soil Use Manage. 31, 29–38.).

The total SIC stock values for 0–0.3 m were relatively similar to those obtained by Marchant et al. (2015). They also mapped topsoil SIC in France and evidenced that "the largest values were recorded in the calcareous and shallow chalky soils of the Champagne region, on the Jurassic rocks of the Charente region (southwest France) and on various calcareous rocks

of the Mediterranean region of southern France." This suggests that the main controlling factors of SIC stocks in French topsoil are the calcareous content of the parent material, its friability, and the rejuvenation of soils, either by deep plowing (vineyards) or erosion (calcareous mountains). Note also that frequent plowing of cropping soils may lead to rock fragmentation and thus to an increase of SIC in fine earth fraction. Indeed, the estimates that we provided are only related to SIC in <2mm fractions. Therefore, large amounts of SIC contained in coarse elements are ignored by these estimates. The total SIC stock may be slightly underestimated for soils developed on dolomitic parent material, as magnesium carbonate is harder to extract with acid than is calcium carbonate. The mean value of SIC where SIC > SOC does not show very large local variabilities. These estimates are limited to the upper 0.5m of soils.

Under the French climate, and without the rejuvenation of calcareous soils, the natural evolution of French soils is progressive acidification. This is why liming is applied on a large proportion of cropped soils. Saby et al. (2017) noted a small increase in pH of French non-calcareous agricultural soils using large datasets on analysis requested by farmers between 1996 and 2010. Though it can be considered beneficial for agronomic and soil structure stability points of view, the amount of lime applied to French soils and their dynamics in topsoil is unlikely to change the global estimates significantly.

Some authors argue that deep weathering of some non-calcareous soils may increase deep soil carbon sequestration when the source of Ca is the weathering of other Ca-bearing minerals (e.g., Cromack et al., 1979; Landeweert et al., 2001). Additionally, amending soils with crushed calcium- and magnesium-rich silicate rocks may lead to an increase of atmospheric CO<sub>2</sub>-derived dissolved inorganic carbon that may then be further sequestered in soils through the formation of carbonate minerals (e.g., Beerling et al., 2020). There is, however, little evidence that this is feasible in the French context.

The total SIC stocks in mainland France topsoil (0-0.3 m) in fine earth are less than three times the total SOC stocks to the same depth. Given the potential of additional carbon sequestration by SOC recently estimated for mainland France (1.8 per mil, Martin et al. (2021)), they are unlikely to contribute to substantial additional C sequestration, and the main issue should be to act against acidification. However, these SIC stocks are somewhat similar to the SOC stocks for the 0.3–0.5 m layer. This is because SOC stocks sharply decline with depth and some chalky and calcareous soils

have very high SIC stocks in this layer. In summary, without deeper analysis, it cannot be stated that France is ready to put in place reliable and efficient actions to increase SIC sequestration in its soils, especially if the overall objective is to mitigate GHG in the atmosphere.

#### 5.8 SIC in the United States of America

The distribution of SIC in the United States is shown in Fig. 17A. The map is based on the STATSGO database of the USDA-NRCS (Ditzler et al., 2017). At the broadest scale, the United States is divided into Land Resource Regions (LRRs). These are units based on a synthesis of physiography, geology, climate, water resources, soils, biological resources, and land use (USDA-NRCS, 2006). The LRR divisions, which are labeled alphabetically from the Northwest to the Southeast, are shown on both the SIC map and Land Use map (Fig. 17A and B). An analysis of the amount of SIC in the LRRs was made based on the minimum, midpoint, and maximum amounts of calcium carbonate equivalent recorded in STATSGO (Guo et al., 2006a). The results show that the greatest amount of SIC occurs in the arid and semiarid LRR-D, LRR-H, LRR-I, and LRR-J (Fig. 18). The glaciated upper Midwestern regions that contain large amounts of detrital lithogenic carbonate also have high amounts of SIC (i.e., LRR-M and LRR-F). Humid regions, such as LRR-A, -N, -O, -P, -R, -S, and -U have little or no SIC. The total SIC for the conterminous USA ranges from 22.6 to 93.7 Pg with a midpoint amount of 54.1 Pg (Guo et al., 2006a).

Pedogenic carbonate is linked to bioclimatic conditions in the United States as shown by progressively deeper carbonates in soils as a function of progressively wetter regional climates (Jenny and Leonard, 1934) and local orographic climates (Gile, 1977). Generally, a mean annual precipitation of roughly 500 mm (20 in.) is the boundary in the United States above which pedogenic carbonates generally do not form (Birkeland, 1999). However, pedogenic carbonates in soils receiving greater than 500 mm do occur in many regions as relict features, i.e., an example of "soil memory" where the carbonates are vestiges of drier climates (Targulian and Goryachkin, 2004).

The SIC-driven threats to agriculture are primarily related to the erosion in croplands with carbonate-rich subsoils (compare Fig. 17A and B). As erosion proceeds and the depth to carbonate-rich below ground becomes progressively shallower, where the availability of phosphorus and various micro-nutrients such as iron extremely decreases in the rooting zone.



**Fig. 17** (A) Soil CaCO<sub>3</sub> distribution in the conterminous United States to a depth of 2 m. Land Resource Regions are shown as black polygons. (B) Land use in the conterminous USA based on 2018 data from the USDA National Agricultural Statistics Service. Figures courtesy of Chad Ferguson, US Natural Resources Conservation Service.



**Fig. 18** Minimum, median, and maximum amount of soil inorganic carbon within the Land Resource Regions of the conterminous United States. *Data from Guo, Y., Amundson, R., Gong, P., Yu, Q., 2006a. Quantity and spatial variability of soil carbon in the conterminous United States. Soil Sci. Soc. Am. J. 70, 590–600.* 

Chlorosis due to iron deficiency is very common in calcareous soils (Havlin, 2014). If the SIC exists as a petrocalcic horizon, i.e., a cemented soil layer due to carbonate precipitation, adverse physical soil properties such as reduced water percolation and aeration prevent root penetration, resulting in reduced crop growth and yield. However, the presence of carbonates, as silt-size particles, may impart favorable water-holding properties to plants (Duniway et al., 2010). Regarding CO<sub>2</sub> storage, it can be hypothesized that the exhumed soil carbonate might be a source of atmospheric  $CO_2$  once the carbonate is elevated out of its subsoil biogeochemical environment and exposed to the topsoil with greater dissolution possibility, due to a higher root density and presence of organic acids. A side-by-side comparison of exhumed vs non-eroded soils with petrocalcic horizons showed no statistical difference in  $CO_2$  emissions, at least in the short-term of 3 years (Serna-Pérez et al., 2006).

With increasing global temperatures, increasing desertification can be expected. Of particular concern is the expansion of desert shrublands into semiarid grasslands (Schlesinger et al., 1990). Based on current observations, both water and wind erosion will increase as bare ground between shrubs increases (Monger and Bestelmeyer, 2006). It has been long recognized that increased water erosion causes siltation of local streams and adversely affects

water sources (Joel et al., 1937). It has also been recognized that increased wind erosion from desert landscapes has broad regional effects on air quality and intercontinental transfer of particles (Yaalon, 1987).

# 6. A review of digital mapping of SIC stocks

As described in the previous section, many regions still rely on traditional mapping approaches for mapping SIC, delineating soil-landscape units based on the interpretation of remote sensing images and geological maps. Digital soil mapping (DSM) approaches were promoted about two decades ago (McBratney et al., 2003), which have been widely adopted to map SOC concentration and stock throughout the world, have improved global estimates of SOC stocks (Arrouays et al., 2014; Chen et al., 2022). However, digital mapping studies specific to SIC are limited and the global SIC stock estimations are still uncertain as such estimates are based on coarse-scale soil information (Eswaran et al., 1993).

Digital soil mapping helps identify regions with the highest threats, priorities, and drivers of change. In addition, digital maps of SOC and SIC can be used in climate models to assess the sensitivity and feedback to future climate change. Based on DSM within the framework of SCORPAN model (McBratney et al., 2003; Minasny and McBratney, 2016) we surveyed studies on the spatial distribution of SIC concerning models, covariates and three-dimensional variations. In a systematic literature survey, the following keywords were searched: "digital soil mapping" and "soil inorganic carbon," "digital soil mapping" and "inorganic carbon," "digital mapping" and "soil carbonate," "digital mapping" and "soil carbonate," "inorganic carbon," AND "map" AND "digital" AND "soil," "carbonate" AND "map" AND "soil."

The search results were then manually filtered to get the specific outcomes on digital mapping of SIC. Only studies related to the modeling and mapping or spatial distribution of SIC were included. Studies on SIC using different spectroscopy techniques were also included.

## 6.1 SIC prediction, mapping and modeling

Digital mapping of SIC follows the procedure of SCORPAN framework (McBratney et al., 2003; Minasny and McBratney, 2016) by relating a

set of soil forming factors as covariates with SIC content as a continuous soil property.

$$SIC_{x,v} = f(S, C, O, R, P, A, N) + e$$
(8)

 $SIC_{x,y}$  represents SIC content at geographical position of x (latitude) and y (longitude) as a function of soil position (S), climate (C), organisms (O), including human effects through land use and management, relief (R), parent materials (P), age or time (A) and spatial position (N). The *e* is the spatially correlated errors. This model links field observations (e.g., carbonate or SIC concentration) to spatial environmental variables captured by proximal or remote sensors via a spatial prediction function *f*. The model assumes the factors are in a steady state, and the observations should cover the entire range of variation in covariates so that the model can be extrapolated to the whole area. The form of *f* can be a simple linear model, regression, or machine learning models (Table 3).

Table 3 shows the summary of the literature review on the digital mapping of SIC. The number of publications for SIC digital mapping is significantly smaller than studies on SOC (Lamichhane et al., 2019; Sharififar and Minasny, 2022). Fig. 19 shows the distribution of SIC digital mapping studies worldwide using DSM. In the following sections, we will discuss the main factors that can help drive more accurate SIC mapping.

*Extent and resolution:* SIC has been mapped from field to continental extents at different spatial resolutions. The majority of the studies have mapped SIC at a resolution of  $\sim$ 30 m. The resolution of the images corresponds to the mapping extent and utility. For example, the 30 m resolution image may only be suited for regional mapping but cannot detect small pockets of SIC. Some national efforts have been made to map SIC (e.g., in Australia, China, France, and United States), however, many countries still lack data or have not mapped SIC.

Where in the world? Fig. 18 shows the publications numbers on SIC digital mapping in different countries. These studies cover various spatial scales from local and regional to national extents. These studies were mainly published in soil science and focused on DSM. In Africa, some parts of Europe and South America, SIC maps are not supplied, possibly because in some of these areas SIC is very low and has not attracted attention as much as in arid and semi-arid regions of the world.

What was mapped? Most of the DSM studies on SIC focused on mapping inorganic carbon concentration or stock. Some studies also mapped SIC

 Table 3 Summary of studies on the digital mapping of soil inorganic carbon.

Study area	Extent	Fitting method	Covariates	Accuracy (R <sup>2</sup> )	Max Soil depth (cm)	Spatial resolution (meter)	No of samples	Uncertainty: confidence of prediction (Yes/No)	References
Australia	4.6 km <sup>2</sup>	Partial least squares regression, regression rule model, zonal statistic for class mapping	Mid infrared spectroscopy, <b>DEM</b> , secondary terrain attributes: <b>EM31 EC</b> <sub>a</sub> , radiometric thorium	RMSE=4.21, Mean observed value=8.92	100	5	60	No	Miklos et al. (2010)
France	National	Partial least square regression and Mid infrared spectroscopy	_	0.97	30	16 km	2086	No	Grinand et al. (2012)
France	24.6 km <sup>2</sup>	Airborne Vis-NIR spectroscopy by PLSR	HyMap hyperspectral image	0.76 (for CaCO <sub>3</sub> ) <sup>a</sup>	5	5	95	No	Gomez et al. (2012)
Turkey	4949.39 km <sup>2</sup>	Landsat 7 ETM+, Linear regression	Short wave infrared band of Landsat 7	~0.71	20	30	164	No	Dogan and Kılıç (2013)
Morocco	15,000 km <sup>2</sup>	Multiple linear regression; X-ray diffraction	ASTER satellite data, material identification and characterization algorithm	0.42 (CaCO <sub>3</sub> )	Not defined	90	73	No	Mulder et al. (2013)
France	550,000 km <sup>2</sup>	Linear mixed model & kriging	Geology(parent material)	(SIC concentration) AIK: 1532.3, prediction error: 0.43	30	2.5 km	~2000	Yes	Marchant et al. (2015)
Australia	National 7.7 Mkm <sup>2</sup>	Decision tree based on piecewise linear regression	Geology, <b>surface</b> <b>geochemistry</b> , soils, climate, vegetation, terrain, multispectral satellite data	0.50	0–10 & 60–80	90	1311 (calcium carbonate)	Yes	Wilford et al. (2015)

Iran	88 km <sup>2</sup>	Artificial neural network, boosted regression tree, generalized linear model, multiple linear regression	<b>Terrain attributes</b> , remote sensing indices, geology map, existing soil map, geomorphology map	0.25 Calcium carbonate equivalent	30	50	120	No	Mosleh et al. (2016)
India	National 3,000,000 km <sup>2</sup>	RF	Remote sensing, climatic, land cover, rock type, soil type, NDVI, irrigation status	0.86	100	250	1198	No	Sreenivas et al. (2016)
Iran	12.25 km <sup>2</sup>	<b>ANN</b> , gene expression programming, multivariate linear regression	Terrain attributes, Landsat 7 images, <b>vegetation features</b>	0.72 (Calcium carbonate equivalent)	10	30	137	No	Mahmoudabadi et al. (2017)
Germany	$3600 \mathrm{km}^2$	Partial least square regression + ordinary kriging for mapping	Diffuse reflectance infrared Fourier transform mid-infrared spectroscopy	Root mean square standardized error=~1	30	200	1170	No	Mirzaeitalarposhti et al. (2017)
China	30,000 km <sup>2</sup>	RF with the help of pedogenic knowledge	Climate data, terrain attributes, Landsat 5 and its derivatives, latitude & longitude positions	LCC: 0.86	100	90	99	No	Yang et al. (2017)
Greece	112.6 km <sup>2</sup>	Artificial neural network	Landsat 8, Sentinel 2, Field proximal sensing	Calcium carbonates: 0.60–0.74	20	30	30	No	Alexakis et al. (2019)
Europe	Continental	Gaussian process regression	MODIS data, DEM, climatic data	Calcium carbonates: 0.61	20	250	21,682	Yes	Ballabio et al. (2019)

Study area	Extent	Fitting method	Covariates	Accuracy (R <sup>2</sup> )	Max Soil depth (cm)	Spatial resolution (meter)	No of samples	Uncertainty: confidence of prediction (Yes/No)	References
USA	150,000 km <sup>2</sup>	Classification and Regression Tree, Bagged Regression Tree, Boosted Regression Tree, <b>RF</b> , <b>Support</b> <b>Vector Machine</b> , Partial Least Square Regression, Regression Kriging, Ordinary Kriging	Most important: lithology & climatic factors	Total carbon: 0.72	20	30	1014	No	Keskin et al. (2019)
Italy	17,200 km <sup>2</sup>	Multinomial logistic regression (for mapping classes of CaCO <sub>3</sub> )	DEM (depth of valleys, flow accumulation), geology, pedoclimate- temperature, soil moisture, morphometric parameters, Landsat 8 images (& spectral bands indices)	Cohen's Kappa = 0.58 (CaCO <sub>3</sub> )	Profile samples; but the exact depth was Not defined	30	1510	No	Piccini et al. (2019)
USA	National	Mid-infrared spectroscopy + partial least squares	_	0.99	-	_	36,000	No	Seybold et al. (2019)
Iran	$860\mathrm{km}^2$	<b>Cubist</b> , random forest, regression tree, multiple linear regression	DEM, Landsat Enhanced Thematic Mapper data; best: ratio vegetation index & Landsat ETM band 4	0.30	30	30	334	No	Zeraatpisheh et al. (2019)
Cypress	9250 km <sup>2</sup>	IDW, Kriging	Parent material as controlling factor	_	70	_	3219	No	Zissimos et al. (2019)

#### Table 3 Summary of studies on the digital mapping of soil inorganic carbon.—cont'd

Iran	567 km <sup>2</sup>	RF and <b>Cubist</b>	Terrain indices and <b>multi-temporal</b> satellite images (Landsat 8)	0.70	30	30	142	Yes	Fathololoumi et al. (2020)
Australia	2650 km <sup>2</sup>	RF, combined model	Land use, NDVI, Topographic data, geology, gamma radiometrics	LCC: 0.40	50	30	154	No	Filippi et al. (2020)
Canada	95 km <sup>2</sup>	Random forest, Cubist, SVM <sup>b</sup> , <b>K-nearest</b> neighbors	Topographic variables, biogeoclimatic subzones, bedrock geology	0.05-0.10	0–145	3	171	Yes	Kasraei et al. (2021)
Russia	0.35 km <sup>2</sup>	Process-based modeling: subordination of soil processes to landscape processes and nesting of the soil system into the landscape	Topography data, climate data, <b>hydrological model</b> (as a landscape process element)	Accuracy <sup>c</sup> : 64%	1-6 m	_	157	Yes	Lozbenev et al. (2021)
Azerbaijan	4.23 km <sup>2</sup>	Universal kriging, Random forest kriging	Terrain attributes, spectral indices, Pl'eiades-1 satellite data	RMSE = 2.00 (CaCO <sub>3</sub> )	15	12.5	115	Yes	Mammadov et al. (2021)
India	89 km <sup>2</sup>	Linear regression	Airborne hyperspectral Vis-NIR spectroscopy	0.58 (CaCO <sub>3</sub> )	Not defined	4	24	Yes	Mitran et al. (2021)
Middle east	3,338,000 km <sup>2</sup>	Random forest (with the help of Google Earth Engine)	30 covariates including: soil, climate, relief, parent material and age features: most important ones: annual precipitation and temperature, elevation	0.54 (CaCO <sub>3</sub> )	20	30	5000	Yes	Poppiel et al. (2021)

<sup>a</sup>Calcium carbonate is considered as inorganic carbon, but might be an underestimation of the whole soil inorganic carbon as it does not include all carbonates. <sup>b</sup>Support vector machine. <sup>c</sup>Accuracy of prediction for Calcium carbonate depth.



**Fig. 19** Number of publications on digital mapping of soil inorganic carbon in different countries within the recent decade. [Only studies on SIC mapping based on DSM framework is shown here].

occurrence probability (e.g., Filippi et al., 2020), or  $CaCO_3$  content (Wilford et al., 2015). Due to the skewness of carbonate distribution, probability of occurrence modeling can help overcome the difficulty of mapping and modeling its spatial variation. The depth to carbonate modeling and mapping could be even more useful for land use planning and soil management, which has not been well explored in the literature.

Inputs: for SIC mapping, most of the studies have used legacy data, often with additional soil sampling. Due to the costs and time limitations, several studies have relied solely on legacy data. In some countries, the soil sampling density was fairly high, although the extent of the study area was small (e.g., Kasraei et al., 2021). In terms of SIC indicators, as explained in the "measuring SIC proximally and remotely" section, different approaches have been used as data input. Using spectral data as inputs for mapping and modeling the spatial distribution of SIC has shown promising predictions, with notably high accuracy of Mid-Infrared (spectral range  $4000-400 \,\mathrm{cm}^{-1}$ ), although not for mapping purposes. As an example, the prediction accuracy of SIC using mid-infrared had an  $R^2=0.97$  in France (Grinand et al., 2012) and  $R^2=0.99$  in the United States (Seybold et al., 2019). In general, using spectroscopy for sensing SIC and as input for further modeling and mapping can be more cost-effective and less time-consuming than conventional laboratory approaches.

*Covariates*: the covariates used for the digital mapping of SIC include various variables that directly or indirectly affect the formation or accumulation of SIC and soil inherent spatial variation; among which, climatic factors and lithological variables have been reported as the most important ones that help model and map SIC. Other covariates comprise digital elevation models (DEM) and terrain attributes derived from DEM. The covariates were mainly obtained via satellite images from sources such as ASTER, Landsat, MODIS and Sentinel. Hyperspectral data from remote sensing have been reported as a successful tool for SIC mapping and prediction (e.g., Gomez et al., 2012; Mitran et al., 2021).

*Modeling approaches*: according to the literature survey, methods of SIC modeling and mapping can be grouped into a few approaches, including:

- (a) the use of digital soil mapping framework with the help of machine learning and/or geostatistical estimation/simulation by applying soil forming factors as explanatory variables (Minasny and McBratney, 2016). The most used algorithms for SIC digital maps have been random forest, multiple linear regression, and some other models, including partial least squares regression, artificial neural networks and Cubist (Table 3).
- **(b)** chemometric modeling—spectroscopy and proximal soil sensing using Visible Near Infrared (0.4–2.5 μm), Mid-Infrared or integrated methods of spectroscopy with the help of proximal or remote sensors (Gomez et al., 2022; Sepahvand et al., 2019). Spectroscopic methods can be advantageous by providing rapid and cost-effective data with desired spatio-temporal resolution.
- (c) pedo-climatic and pedogenic knowledge-led mapping; a helpful SIC mapping approach using natural soil and biogeoclimatic characteristics (such as pedocal concept (Marbut, 1935; Salley et al., 2016)) for zoning areas with homogeneous mapping units or sub-zones (Bui et al., 2009; Yang et al., 2017).
- (d) the use of so-called taxotransfer rules, which estimate soil properties based on modal characteristics of soil units/taxons (Batjes, 2016).

Unlike SOC, SIC can be difficult to model and map in some areas, as SIC is not contiguously distributed in an area. The measured datasets are often zero-inflated, meaning that many locations have low or no SIC content. To overcome this problem, SIC occurrence can be first predicted, then the stock can be quantified. For example, Filippi et al. (2020) presented such an approach, where in this case the presence of SIC in the study area was modeled using a random forest model, then used a separate model developed to predict amounts of SIC in those areas where SIC was predicted to be present. SIC change: Unlike SOC, mapping SIC changes has not been widely explored, possibly because SIC is normally not expected to show notable changes within time. Song et al. (2022) mapped SIC stock of China using data from a national survey in the 1980s and compared it with data from 2010s. They found an overall decrease in SIC stocks in the top 30 cm (0.11 t Cha<sup>-1</sup> year<sup>-1</sup>). Total SIC stocks have decreased by 9% ( $1.37\pm0.37$  Pg C). In a semi-arid field in the great plains of the United States, Sherrod et al. (2015) hypothesized that CaCO<sub>3</sub> concentration in the soil surface layer is inversely correlated with the change in surface elevation due to erosion and deposition. Using high-resolution DEM, the temporal differences in elevation were related to the change in CaCO<sub>3</sub> concentration. This idea should be more explored as a tool for investigating SIC changes.

Validation and uncertainty of SIC maps: the validation approach that most DSM studies have done is based on the data hold-back method, in which a portion of collected data (usually 20–30% of the whole dataset) is used for testing the correctness of the predicted pixels of the produced map. The "leave-one-out" method can be an efficient alternative when the dataset is small. However, the K-fold cross-validation method can help get a more robust accuracy outcome for skewed data, typical for SIC. In terms of uncertainty quantification, only about 35% of the studies have carried out uncertainty assessment of the digital maps of SIC. There is no report of a specially designed mapping study for SIC along with a designed validation survey—a clear need for the future.

#### 6.2 Laboratory analysis, proximal and remote sensing

There are several methods of measuring and detecting the existence of SIC, such as HCl test for carbonates. In the laboratory, the standard methods include the elementar analyzer (i.e., the difference between the measured total C content before and after acidifying the soil sample), calcimetery (i.e., measuring the pressure of  $CO_2$  gas produced following sample acidification), and the titration method to determine the total neutralizing value (TNV) also known as calcium carbonate equivalent (CCE) (Loeppert and Suarez, 1996).

It is important to know the appropriate methods of SIC evaluation, as different soils might require specific approaches (Apesteguia et al., 2018). For example, using calcimetery when different carbonate types are present in soil may lead to underestimating SIC content, especially when only calcium carbonate content is reported as the SIC indicator. Methods for soil

carbon measurement with the relative advantages and disadvantages are discussed in detail in Nayak et al. (2019) and Nelson and Sommers (1996).

Indirect measurement using infrared reflectance of soil: Mid-Infrared (MIRS, 2500-25,000 nm, 4000-400 cm<sup>-1</sup>) and Visible Near Infrared (Vis-NIR,  $400-2500 \,\mathrm{nm}$ ,  $25,000-4000 \,\mathrm{cm}^{-1}$ ) reflectance spectroscopy have shown to be able to quantify SIC (Gaffey, 1986; Wijewardane et al., 2018). MIR spectroscopy is based on absorption bands corresponding to fundamental molecular vibrations, while Vis-NIR spectroscopy is based on absorption bands corresponding to weak overtones and combinations of fundamental vibrations (Williams, 1987). Strong fundamental vibrational states related to soil components occur in the MIR between wavelengths 2500 and 25,000 nm, including the ones at 700, 880, 1450 and 2510 cm<sup>-1</sup> related to carbonates (Legodi et al., 2001) (Fig. 20A). Several additional bands attributed to carbonates can also be identified at 3000-2900 cm<sup>-1</sup> due to overtones, and at 2600-2500 and 1830-1760 cm<sup>-1</sup>, due to combinations of fundamental vibrations (Comstock et al., 2019; Legodi et al., 2001; Nguyen et al., 1991) (Fig. 20A). Overtones and combination modes overlap in the Vis-NIR spectral range, and carbonates have absorption bands at 1.6,



**Fig. 20** Examples of absorbance spectra of four soil samples collected in Tunisia with varying SIC content using laboratory (A) MIR and (B) Vis-NIR spectroscopy.

1.7, 2.2, 2.3, and  $2.5\,\mu\text{m}$  (Ben-Dor and Banin, 1995) (Fig. 20B). As bands are not distinct, estimating carbonate using Vis-NIR is more challenging than MIR.

Several methodologies have been successfully tested to link MIR and Vis-NIR soil spectra to SIC content, such as simple linear regression (SLR) using specific absorbance peaks (Legodi et al., 2001), using the MIR and Vis-NIR domain, respectively, and partial least-squares regression (PLSR) using the entire spectra (Gomez et al., 2022; Wijewardane et al., 2018). Machine learning algorithms such as artificial neural networks, support vector machines, Cubist regression trees, random forest, and deep learning have been successfully applied for estimating SIC concentration (e.g., Dangal et al., 2019; Gruszczyński and Gruszczyński, 2022; Ng et al., 2022).

In situ approaches for measuring SIC have also been proposed using the Vis-NIR spectrometer, laser induced breakdown spectrometer and inelastic neutron scattering method (Chatterjee et al., 2009). Usually, in situ and ex situ measurement methods can result in somewhat different accuracies. However, the accuracies of these methods can come close to each other when suitable treatments are applied on ex situ methods, as sometimes commonly used for spectroscopic measurements (Leogrande et al., 2021; Wang et al., 2012).

While laboratory Vis-NIR soil spectroscopy may provide correct SIC estimations, Vis-NIR imaging sensors may provide lower accuracies of SIC estimations as these sensors are subject to degradations due to the atmospheric effects, Signal-to-Noise Ratio of sensors, coarse spatial and spectral resolutions, and soil surface conditions (e.g., Chabrillat et al., 2019; Gomez et al., 2018; Lagacherie et al., 2008). For example, Vaudour et al. (2019) reported  $R^2$  from 0.15 to 0.48 for predicting CaCO<sub>3</sub> using Sentinel-2 satellite images in Mediterranean and Temperate contexts. Additionally, Gomez et al. (2008a,b) and Lagacherie et al. (2008) reported  $R^2$  of 0.77 and 0.61, respectively, for SIC prediction based on Vis-NIR hyperspectral imaging sensors (AISA-DUAL and Hymap sensors, respectively).

### 6.3 Summary of DSM review

The review of DSM studies shows that SIC can be successfully mapped using the DSM approach, despite the modest number of studies. Some notable gaps in current SIC mapping are:

• The shortage of DSM studies and ignored importance of SIC

Most of the efforts in soil C studies have focused on organic carbon, in terms of its up-to-date data collection, monitoring and mapping. Many countries do not have any updated SIC database, nor do they have a regional or national extent map for it. Furthermore, in the global scale efforts, SOC global maps have been updated in recent years with reasonable spatial resolution, while there is no accurate and updated global map for SIC.

Challenges in mapping SIC

The DSM of SOC has rapidly progressed worldwide due to the interest in soil carbon sequestration; however, a modest effort was found for DSM of SIC. This could be due to the lack of interest in SIC, but mapping of SIC is more challenging. Many studies successfully mapped topsoil SOC using remote sensing images and other covariates, however, as SIC mostly presents in the deeper depths, covariates that mainly represent surface conditions cannot detect SIC. That also means surveys for SIC needs to sample and measure the whole soil profile. In addition, in some areas, SIC does not exist contiguously, and thus more extensive surveys is needed to first establish its extent.

The lack of uncertainty assessment of SIC maps

Using the machine-learning algorithms, with the help of "random sampling with/without replacement" and bootstrapping, confidence intervals of predictions in each pixel of SIC map can be evaluated. The literature review showed the lack of exploring stochastic models rather than deterministic models for predicting, modeling and mapping SIC stocks, its depths of accumulation, probability of occurrence and mapping. The application of simulation and stochastic modeling might pave the way for addressing the difficulty in mapping the distribution of highly skewed SIC data. Simulation models can be advantageous as to higher consistency of their outputs, compared to machine learning bootstraps as an alternative uncertainty quantification.

Monitoring and mapping SIC stock change

While it was thought before that SIC stock is stable over time, there is some evidence that SIC loss can happen within decades, due to anthropogenic interventions. This can affect the sustainability of agricultural production and be more of a significant contributor to climate change than previously thought. Hence, monitoring SIC change is more necessary than once thought before. Multi-source covariates provide geographical information, elevation and optical and radar images which provide better information than a single covariate and can be useful tools for monitoring and mapping SIC temporal changes. Such covariates can help with a better spatial and temporal resolution for monitoring any probable change in SIC stock, mainly in topsoil. As SIC is mainly present in the subsoil, proximal sensors that can penetrate deeper into the soil profile are required for a more detailed assessment.

# 7. SIC and SOC relationships

The SIC and SOC are usually considered and mapped separately, while SIC may also control the dynamics of SOC and vice versa. SIC can control SOC by influencing pH and thereby microbial activity,  $Ca^{2+}$ -SOC binding and by aggregate formation and stability. Although the direct or indirect role of SIC and  $Ca^{2+}$  binding on SOC stabilization is not entirely well-defined (Rowley et al., 2018), the precipitation of pedogenic carbonates can be a carbon-protecting action, as it can cement aggregates and thus enhance soil aggregation. Three main mechanisms have been suggested to explain the positive impact of SIC on SOC stabilization: (i) aggregation and the occlusion of SOC, (ii) organo-mineral and organo-cation interactions, and (iii) inclusion of SOC within pedogenic or biogenic  $CaCO_3$  (Rowley et al., 2018). These mechanisms are mostly linked to the reactivity of the  $Ca^{2+}$ , favoring soil aggregation.

Some studies suggested significant protection of SOC within macroaggregates in calcareous soils in the Mediterranean region (Fernández-Ugalde et al., 2011; Martí-Roura et al., 2019). The structure of macro-aggregates in calcareous soils can be different from carbonate-free soils. These aggregates hold the active precipitation of carbonates (Catoni et al., 2012; Falsone et al., 2010) and favor the stability of SOC. Aggregation can protect SOC from decomposition and favor SOC accumulation and stabilization (Balesdent et al., 2000). The degree of SOC protection inside aggregates of calcareous soils is yet to be quantified. In addition, some forms of SIC may also alter SOC decay rates by forming physical coatings around particles or enhanced aggregation (Zornoza et al., 2018).

Conversely, SOC can influence SIC. Carbon for the formation of pedogenic carbonates is generally supplied from  $CO_2$  in the soil solution, because of respiration and decomposition of soil organic matter (Hasinger et al., 2015; Zamanian et al., 2016), demonstrating an interactive relationship between SOC and SIC. Anthropogenic factors can greatly influence this relationship and change the balance between SIC and SOC.

For example, liming increases pH and decreases fungal diversity but may increase bacterial diversity leading to SOC mineralization (Xiao et al., 2018).

SIC shows higher spatial variability than SOC, even within given land use types (Stevenson et al., 2005; Wang et al., 2018). The transition from desert to shrublands and crop lands has shown a higher accumulation of both SOC (usually in topsoil) and SIC (mostly in subsoil) (Wang et al., 2018). Positive correlations between SIC and SOC have been reported (Guo et al., 2016; Wang et al., 2015; Yang et al., 2021a,b; Zhang et al., 2010). The positive relationship has been mostly reported for croplands and in agricultural practices where soil organic matter contributes to SIC formation by providing carbon source for pedogenic carbonate (Zamanian et al., 2016). On the other hand, a negative relationship between SOC and SIC has been reported for non-agricultural areas (Lu et al., 2020; Raheb et al., 2017; Sharififar et al., 2019a). In such areas, erosion, deposition, and the redistribution of soils are considered the dominant factors affecting SIC and SOC relationship. In addition, biotic and abiotic processes and agricultural practices (e.g., fertilization and liming) can affect the relationship between SIC and SOC, and their abundance at different soil depths.

When assessing the SOC and SIC relationship, land use interaction with pedogenic and geologic phenomena should be considered. Hence, more comprehensive studies with wide-ranging sampling efforts across different soils will help deepen our understanding of the SOC and SIC relationship and balance. This will help find the probable threshold at which SOC and SIC correlation changes in different climatic conditions and land uses. This would also enhance our ability to model a complete picture of soil C and its relation with climate change.

Important factors that need to be considered for SIC and SOC balance and total soil C mapping and modeling are land use change and anthropogenic factors that might change the thresholds where the SIC and SOC relationship changes. As shown in Fig. 7, the difference in SIC and SOC stocks is more pronounced in the northern hemisphere than more southerly latitudes. If we can understand the spatial correlation between SIC and SOC, then SOC data/maps (which are much more available than SIC maps) can be used as covariates for SIC mapping.

## 8. Human influence and climate-change effects on SIC

There is a general belief that SIC has a millennial residence time and thus has a negligible contribution to global C cycle compared to SOC (Schlesinger, 1985; Zamanian and Kuzyakov, 2022; Zamanian et al., 2016). Therefore, the SIC potential as a regulator of atmospheric  $CO_2$  concentration over human lifetime scales has not been fully explored, especially relative to our understanding of the role of SOC. Nevertheless, SIC, particularly in arid and semiarid regions of the world, can be the main soil C pool, significantly contributing to C sequestration. For example, the amount of sequestered C as SIC in the first and second meter of soils in the conterminous USA, depending on vegetation, i.e., shrublands, grasslands and pasture, is about 1.27 to twice the amount of sequestered C as SOC (Guo et al., 2006a,b).

Moreover, the formation of pedogenic carbonates due to the significant contribution of silicate weathering in Canadian prairies, resulted in 1.4 times more C sequestration as SIC than SOC (Landi et al., 2003). In fact, SIC can act as a net sink of atmospheric  $CO_2$  (Lapenis et al., 2008; Laudicina et al., 2013) and help reduce the environmental costs of carbon loss and  $CO_2$  emission across farms (Groshans et al., 2019).

Several natural and anthropogenic factors control the contribution of SIC in C sequestration or  $CO_2$  emission. Planting trees and shrubs, especially in arid and semi-arid regions or on unproductive saline and alkaline soils can boost  $CO_2$  in the soil through deeper roots. This will increase  $CO_2$  partial pressure in soil, which may directly result in C sequestration as the H<sub>2</sub>CO<sub>3</sub> ion species in soil (Carmi et al., 2019) or lead to silicate weathering, Ca<sup>2+</sup> release and pedogenic carbonate formation (Haque et al., 2020).

However, there is increasing evidence that the SIC has contributed to  $CO_2$  emissions through human activities such as management practices, especially irrigation, liming and incorporation of legumes into cropping systems (Lal et al., 1999; Shi et al., 2012). Recent studies indicate that the SIC pool progressively becomes more sensitive to anthropogenic acidification and acts as a  $CO_2$  source (Martin, 2017; Monger et al., 2015; Tao et al., 2022; Zamanian et al., 2018, 2021). Intensive agriculture with ammonium-based fertilization results in soil acidification that accelerates the SIC loss and makes SIC a net source of atmospheric  $CO_2$  (Zamanian et al., 2018).

Irrigation can increase or decrease SIC stocks depending on water amount, water quality and soil conditions. For example, irrigation water can supply  $Ca^{2+}$  (and  $Mg^{2+}$ ) ions to the soil (Bughio et al., 2016), which may subsequently lead to the formation of pedogenic carbonates (de Soto et al., 2017, 2019). The fate of the leached bicarbonate ions through the soil profile can also have different consequences for the C cycle. If the leached bicarbonate ions are re-precipitated at lower soil depths, the process will be carbon neutral. It is still carbon neutral if the carbonate is leached into the groundwater and ends up precipitated in the oceans.

Liming acid soils can have direct and indirect effects on soil carbon. Liming can directly improve acid soils and sequester  $CO_2$  by increasing SOC (Hamilton et al., 2007). Liming can also indirectly affect soil C protection by preserving organic C in soil via, for example, calcium bonding (Eze et al., 2018; Fornara et al., 2011).

As SOC decay and SIC dissolution/precipitation are affected by a complex suite of factors, the global C balance in agricultural soils rich in SIC is still uncertain (Paradelo et al., 2015; Wu et al., 2009). A few studies have tried to understand and explain the contradiction between the results of the impact of soil management on SIC dynamics (Monger et al., 2015).

Climate and hydrological cycles (i.e., precipitation and evapotranspiration) at local to global scales strongly influence inorganic carbon dynamics and their contribution to atmospheric  $CO_2$  production (Lapenis et al., 2008). For example, scarce rainfall will decrease the dissolution of CaCO<sub>3</sub> and the leaching of dissolved Ca<sup>2+</sup> and Mg<sup>2+</sup>, which may result in more precipitation of SIC than losses via leaching. The amount of local rainfall determines the depth of the wetting front in the soil profile, coinciding with the presence of SIC in the profile. The rainwater quality can affect the inputs of Ca<sup>2+</sup> and Mg<sup>2+</sup> ions to the soil, hence controlling the precipitation of SIC (Wilford et al., 2015). In general, the precipitation to potential evapotranspiration ratio determines SIC dynamics (Tao et al., 2022). Such linkages between the alterations in SIC stocks and climate change further emphasize the importance of DSM in producing an accurate distribution of SIC.

With the CO<sub>2</sub> levels expected to increase to 950 ppm by 2099 (Allen et al., 2018), there is an expectation of increased plant biomass and rhizosphere activity, which could increase soil acidity due to respiration and release of organic acids. Increasing soil acidity tends to dissolve CaCO<sub>3</sub> and leads to the leaching of Ca<sup>2+</sup> and Mg<sup>2+</sup> ions (Kuzyakov et al., 2019; Raza et al., 2021). An increase in air and soil temperature also increases organismal activity (respiration and release of organic acids), which would intensify the release of CO<sub>2</sub> from carbonate (Ahmad et al., 2015; Chevallier et al., 2016). On the other hand, SIC dissolution by organic acid and respiration can lead to CO<sub>2</sub> sequestration (Appendix; Eq. A1). The threshold in which such interactions could lead to net zero emission is still

not well investigated. Furthermore, the reactions in the field are more complex than the theories. When SIC is precipitated in deeper depths, the released  $CO_2$  does not necessarily return to the atmosphere but contributes to soil  $CO_2$  (Carmi et al., 2019), which can lead to the dissolution of minerals such as wollastonite (CaSiO<sub>3</sub>), where its weathering can increase SIC accumulation in soil and  $CO_2$  sequestration (Haque et al., 2020).

Globally, SIC and SOC are being lost due to anthropogenic activity. An estimate of annual losses of SOC due to land cover change is around 1.93 Pg C (Padarian et al., 2021), while estimated annual SIC loss due to fertilization is about 7.5 Tg C (Zamanian et al., 2021). The potential of SIC as a source of  $CO_2$  is even larger when we consider the contribution of liming. Neutralization of agricultural lime, i.e., a mixture of  $CaCO_3$  and  $MgCO_3$ , which is added to acid soils to optimize the pH value for crop production, further releases 273 Tg C per year (Zamanian et al., 2021).

With climate change, alterations in soil C stocks become more important. While SOC stocks can be increased through better management, SIC losses are irreversible, with no expected equilibrium as long as fertilization is applied (Raza et al., 2021). Furthermore, SOC stability, at least in calcareous soils, can be controlled by SIC (Ahmad et al., 2015; Fornara et al., 2011). Nevertheless, the dynamics of SIC because of future climate change and land management are generally overlooked. Current climate-change scenarios indicate that about 7% of Australian land will experience increased drying until 2050. About 6.5 million km<sup>2</sup> are foreseen to suffer from 10% decline in surface soil moisture in the driest decade of 2030 to 2040 (Guglielmo et al., 2021). Despite potential consequences of such climatic changes for alteration in SIC stocks, they have been so far unaccounted for predicting the dynamics of SIC. Because the focus has almost always been to study the dynamics of SOC in response to climate change and global warming. Similarly, the current and future land-use changes that result in acidification and/or changes in soil hydrology will also affect the persistence of SIC and its losses. Thus, global efforts to mitigate climate change through land management and soil C sequestration should also consider SIC.

# 9. Concluding remarks

Soil carbon is regarded as a key negative emission technology. This C sequestration potential of soils will be compromised if SIC is lost through

agricultural practices such as irrigation and fertilization, which are usually employed for organic C sequestration. We need to account for both SOC and SIC to build a complete picture of the soil C cycle. In addition, understanding the interaction of both C forms helps build resilient soils to global changes. Currently, the spatial information on SIC distribution and stock is relatively scarce and digital soil mapping efforts to address this are modest. Furthermore, we do not have a complete joined-up soil C model that explicitly accounts for all sources and sinks of soil carbon.

This review showed that many aspects of SIC in DSM and soil C studies have been so far ignored and that SIC has a crucial role in climate regulation. SIC and SOC need to be integrated into DSM and integrate SIC into a SOC modeling framework. With the help of new mechanistic models that consider both SIC and SOC, we can comprehensively monitor and predict soil carbon changes over decadal scales and at local to global spatial scales. This is of high importance to secure our soil resources against the negative impacts of global changes.

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## Appendix

### CO<sub>2</sub> Emission and Sequestration Reactions

To understand the net contribution of SIC in atmospheric CO<sub>2</sub>, the fate of the dissolved Ca ions should be considered. The incomplete leaching results in the accumulation of dissolved Ca ions in lower soil depths, which could lead to  $CaCO_3$  re-precipitation and  $CO_2$  emission. This reaction frequently occurs in arid and semi-arid regions, where evapotranspiration is higher than precipitation and roots take water up from deeper depths. Thus, under

natural conditions (Eq. A1), CaCO<sub>3</sub> dissolution and re-precipitation will be neither a sink nor a source of atmospheric CO<sub>2</sub> (Zamanian et al., 2018), although, it provides a temporary sink of C as long as HCO<sub>3</sub><sup>-</sup> stays in solution (Eq. A1) (Monger et al., 2015; Zamanian et al., 2021). CaCO<sub>3</sub> dissolution leads to C sequestration (Eq. A1 goes to right), while CaCO<sub>3</sub> re-precipitation results in CO<sub>2</sub> release (Eq. A1 goes to the left) (Zamanian and Kuzyakov, 2022):

$$CaCO_3 + CO_2 + H_2O \leftrightarrow Ca^{2+} + HCO_3^{-} + HCO_3^{-}$$
(A1)

(C in red is carbonate sourced C, and C in blue is CO<sub>2</sub> sourced C)

It should be noted that lithology and water availability can determine the amount of C flux from soils (Hartmann, 2009). In this regard, the rock and mineral type matter, as silicate minerals can have higher potential to sequester atmospheric CO<sub>2</sub> (Eqs. A2 and A3), compared to carbonate minerals. The following reaction of olivine shows the capacity of silicates as a net sink for up to 50% of the consumed atmospheric CO<sub>2</sub>:

$$Mg_{2}SiO4 + 4CO_{2} + 4H_{2}O \rightarrow 2Mg^{2+} + 4HCO_{3} + H_{4}SiO_{4} \rightarrow 2MgCO_{3}\downarrow + SiO_{2}\downarrow + 2CO_{2}\uparrow + 2H_{2}O \quad (A2)$$

For albite, the net sink will be 100% of the consumed atmospheric CO<sub>2</sub>:

$$2NaAISi_{3}O_{8} + 2CO_{2} + 11H_{2}O \rightarrow Al_{2}Si_{2}O_{5}(OH)_{4} + 2Na^{+} + 2HCO_{3}^{-} + 4H_{4}SiO_{4} \rightarrow 2Na^{+} + 2HCO_{3}^{-} + SiO_{2}\downarrow$$
(A3)

Weathering of silicates (e.g., as in Eq. A4) may lead to the release of Ca and C sequestration via CaCO<sub>3</sub> precipitation (Cueva et al., 2019; Mikhailova and Post, 2006; Wang et al., 2014).

$$\operatorname{CaSiO}_{3}(s) + 2\mathbb{CO}_{2} + 3\operatorname{H}_{2}\operatorname{O} \rightarrow \operatorname{Ca}^{2+}(aq) + 2\operatorname{HCO}_{3}(aq) + \operatorname{H}_{4}\operatorname{SiO}_{4} \rightarrow \operatorname{CaCO}_{3}(s) + \operatorname{CO}_{2}\uparrow + 3\operatorname{H}_{2}\operatorname{O} + \operatorname{SiO}_{2} \quad (A4)$$

Weathering of silicates is generally considered a carbon sequestering process (Eq. A4) (Cipolla et al., 2021a,b). While, anthropogenic interventions such as fertilization of calcareous soils (Eq. A5) and liming of acid soils (Eq. A6) lead to  $CO_2$  emission out of SIC (Azeem et al., 2022; Raza et al., 2021; Xie et al., 2021; Zamanian and Kuzyakov, 2019; Zamanian et al., 2018).

For example, the following reactions of urea  $(CO(NH_2)_2)$  shows the release of  $CO_2$ . Note that, nitrate is kept in the formula to show the source of acidity:

$$CO(NH_2)_2 + H_2O \rightarrow 2NH_3 + CO_2\uparrow$$

$$NH_3 + H_2O \rightarrow NH_{4^+} + OH^-$$

$$NH_{4^+} + OH^- + 2O_2 \rightarrow NO_3^- + H^+ + 2H_2O$$

$$CaCO_3 + 2NO_3^- + 2H^+ \rightarrow Ca^{2+} + 2NO_3^- + CO_2\uparrow + H_2O$$
(A5)

And the effect of liming is illustrated in the following:

$$CaCO_3 + H_2 - X \rightarrow Ca - X + H_2O + CO_2\uparrow$$
 (A6)

where, X is an exchangeable site such as soil clay particles.

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